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Tetrahedron

Tetrahedron 63 (2007) 7624-7633

Enantioselective syntheses of (-)-pinellic acid, α - and β -dimorphecolic acid

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Received 6 February 2007; revised 26 April 2007; accepted 10 May 2007 Available online 17 May 2007

Abstract—An efficient enantioselective convergent approach for the synthesis of (–)-pinellic acid 1, α - and β -dimorphecolic acid (2 and 3) from 1,9-nonane diol is described. The synthetic strategy features Sharpless asymmetric hydroxylation, Sonogashira coupling and Birch reduction.

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1. Introduction

Influenza is an infectious disease that infects birds and mammals (primarily in the upper airways and lungs in mammals) and is caused by an RNA virus of the *Orthomyxoviridae* family (the influenza viruses). The most common and characteristic symptoms of influenza in humans are fever, pharyngitis, myalgia, severe headache, coughing, and malaise. Flu rapidly spreads around the world in seasonal epidemics,¹ killing millions of people in pandemic years, and hundreds of thousands in nonpandemic years.

Kampo medicine, 'Sho-seiryu-to' was found to possess potent adjuvant activity by oral administration on nasal influenza infection and nasal influenza vaccination.² (–)-Pinellic acid **1** (9*S*,12*S*,13*S*-trihydroxy-10*E*-octadecenoic acid, Fig. 1) was isolated from the tubers of *Pinellia ternata*, one of the eight component herbs of the Kampo formula, Sho-seiryu-to (SST).³ Pinellic acid is a novel and potentially useful oral adjuvant when used in conjunction with intranasal inoculation of influenza HA vaccines.^{3a} Pinellic acid showed no haemolytic activity.^{3a} Among the series of pinellic acid isomers, the (9*S*,12*S*,13*S*)-compound, which is a natural product, exhibited the most potent adjuvant activity.⁴

Unsaturated hydroxy fatty acids (2–5) play an important role in biological systems and were isolated from both animals and plants. β -Dimorphecolic acid (2) is a unique hydroxydienoid fatty acid, which was first isolated from the seed oil of *Dimorphotheca aurantiaca*.⁵ It was also isolated from *Osteospermum aurantiaca*⁵ and *Osteospermum ecklonis*.⁶

However, its diene congener α -dimorphecolic acid 3 was isolated from the plant Glechoma hederacea L. Labiatae⁷ (commonly known as 'lierre terrestre', 'ground ivy' or 'creeping Charlie'), which has been demonstrated to be a calcium specific ionophore,⁸ an inhibitor of acetylcholine esterase (ACE)⁹ and aromatase,¹⁰ as well as being implicated in the pathogenesis of familial Mediterranean fever.¹¹ In addition, it can competitively inhibit enzyme activity stimulated by PGE₁ and PGD₂. This fatty acid has been previously isolated from Tragopogon porrifolius L. Compositae¹² and Xeranthemum annuurn Asso Compositae.¹³ It has also been found in rice plants infected with blast disease¹⁴ and in bovine heart mitochondria.⁸ This fatty acid contains structural features that are not typically found in plant fatty acids, including a C-9 hydroxyl group and C10-C12 conjugated double bonds, which possess a wealth of biological properties.¹⁵ Owing to their lipid nature, longchain fatty acids play a vital role in maintaining cellular properties,¹⁶ and consequently can elicit a variety of biological responses. Little is known about the biological properties of β -dimorphecolic acid.

In the interest of evaluating the biological and pharmacological properties of these compounds, it was necessary to obtain sufficient quantities by chemical synthesis. Ōmura et al. reported the first synthesis of (–)-pinellic acid 1 and determined its absolute configuration by synthesizing all its isomers via regioselective Sharpless asymmetric dihydroxylation and stereoselective reduction,^{17a,b} followed by another recent synthesis by Sabitha et al. via Sharpless asymmetric epoxidation.^{17c} Rama Rao and co-workers^{18a} accomplished the first total synthesis of racemic dimorphecolic acid and coriolic acid **4** from tetrahydrofurfuryl chloride. Subsequently Takeda and co-workers^{18b} also synthesized racemic dimorphecolic acid by using Grignard

Keywords: Asymmetric dihydroxylation; Sonogashira coupling; Influenza.

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Figure 1.

reactions. While Sato and co-workers^{19a} accomplished stereospecific synthesis of α -dimorphecolic acid by employing Cu^I–Pd⁰ coupling as a key step, the enantioselective synthesis of β -dimorphecolic acid **2** was reported by Ley and Meek^{19b} by utilizing a π -allyltricarbonyliron lactone complex to control the formation of all the stereochemical features of the natural product. As a part of our research programme aimed at developing enantioselective synthesis of naturally occurring lactones²⁰ and amino alcohols,²¹ we have accomplished the stereoselective total synthesis of (–)-pinellic acid **1**, α - and β -dimorphecolic acid (**2** and **3**) starting from the commercially available starting material 1,9-nonane diol employing the Sharpless asymmetric dihydroxylation as the source of chirality.

2. Results and discussion

Our retrosynthetic strategy for the synthesis of (–)-pinellic acid 1, α - and β -dimorphecolic acid (2 and 3) is outlined in Scheme 1. We envisioned that the 12*S*,13*S syn*-diol 17 could be prepared from 1,3-enyne 16, which in turn would be prepared by Sonogashira coupling²² of chiral propargylic alcohol 13 with vinyl iodide 15. Chiral propargylic alcohol 13 could be prepared from 1,9-nonane diol 6, through base induced elimination of chloro compound 12. In this strategy, the 9*S* hydroxy group could be installed through Sharpless asymmetric dihydroxylation²³ of an olefin 8, which in turn would be prepared from 1,9-nonane diol 6.

2.1. Synthesis of chiral propargylic alcohol

The synthesis of (-)-pinellic acid 1, α - and β -dimorphecolic acid (2 and 3) started from commercially available 1,9-nonane diol 6 as illustrated in Schemes 2 and 3. Thus, selective



Scheme 2. Reagents and conditions: (a) p-CH₃OC₆H₄CH₂Br, NaH, dry DMF, cat. TBAI, 0 °C to rt, 1 h, 95%; (b) (i) (COCl)₂, DMSO, Et₃N, CH₂Cl₂, -78 °C to -60 °C; (ii) Ph₃P=CHCO₂Et, benzene, reflux, 4 h, 91%; (c) (DHQ)₂PHAL, K₂CO₃, K₃Fe(CN)₆, MeSO₂NH₂, OsO₄ (0.1 M solution in toluene), *t*-BuOH/H₂O (1:1), 0 °C, 24 h, 96%; (d) *p*-TSA, 2,2-DMP, CH₂Cl₂, overnight, 94%; (e) DIBAL-H, CH₂Cl₂, 0 °C to rt, 2 h, 95%; (f) *N*-chlorosuccinimide, PPh₃, CH₂Cl₂, 0 °C to rt, 3 h, 89%; (g) *n*-BuLi, HMPA, THF, -42 °C to rt, 30 min, 82%; (h) TBDPSCl, imidazole, CH₂Cl₂, 0 °C to rt, overnight, 98%.



Scheme 1. Retrosynthetic analysis of (-)-pinellic acid (1), α - and β -dimorphecolic acid (2 and 3).



Scheme 3. Reagents and conditions: (a) $Pd(PPh_3)_2Cl_2$, CuI, Et₃N, 6 h, 86%; (b) $(DHQ)_2PHAL$, K_2CO_3 , $K_3Fe(CN)_6$, $MeSO_2NH_2$, OsO_4 (0.1 M solution in toluene), *t*-BuOH/H₂O 1:1, 0 °C, 24 h, 91%; (c) Na/liq NH₃, THF, -40 °C, 90%; (d) *p*-TSA, 2,2-DMP, CH₂Cl₂, overnight, 92%; (e) (i) IBX, EtOAc, reflux, 5 h; (ii) NaClO₂, DMSO, NaH₂PO₄, rt, overnight, 81%; (f) HCl (cat.), MeOH, rt, overnight, 78%.

mono protection of 6 with p-methoxybenzyl bromide in the presence of NaH gave the monoprotected diol 7 in 95% yield, which was oxidized to the corresponding aldehyde under Swern conditions,²⁴ and subsequently treated with (ethoxycarbonylmethylene)-triphenylphosphorane in benzene at reflux to furnish the *trans*-olefin 8 in 91% vield. Olefin 8 was treated with osmium tetroxide and potassium ferricyanide as co-oxidant in the presence of (DHQ)₂PHAL ligand under asymmetric dihydroxylation (AD) conditions²³ to give the diol 9 in 96% yield with 99% ee. Treatment of diol 9 with 2,2-dimethoxypropane in the presence of p-TSA gave compound 10, which on reduction with DIBAL-H furnished the alcohol 11 in excellent yield. The alcohol 11 was converted to chloride 12 in 89% yield. Propargylic alcohol 13 was obtained by treatment of 12 with *n*-BuLi in the presence of HMPA²⁵ in 82% yield. The free hydroxy group of 13 was protected with TBDPSCl to furnish compound 14.

2.2. Synthesis of (-)-pinellic acid (1)

Having completed the synthesis of chiral propargylic alcohol 14, our next aim was to prepare envne 16 through Sonogashira coupling²² and carry out subsequent transformations in order to get the target molecule 1. Towards this, the coupling of 14 with *trans*-vinyl iodide 15^{26} using Pd(PPh₃)₂Cl₂ and CuI in triethylamine furnished the 1,3envne product 16 in excellent yield (Scheme 3). Asymmetric dihydroxylation²³ of 1,3-enyne **16** under standard conditions gave the acetylenic diol 17 in good yield with high diastereomeric excess (de >96%) as judged by ¹H and ¹³C NMR spectral analysis. Reduction of alkyne 17 to the (E)-alkene and concomitant removal of the PMB group proceeded smoothly under Birch conditions using Na/liq NH₃²⁷ to afford 18 in 90% yield. Diol 18 was protected as its isopropylidene derivative 19 in the presence of 2,2-dimethoxypropane and a substoichiometric amount of p-TSA in good yield.

Oxidation of the primary alcohol in **19** to the corresponding aldehyde using IBX^{28} and further oxidation using $NaClO_2$ in DMSO under buffered conditions²⁹ furnished the acid **20**. Finally, the acetonide and TBDPS groups were deprotected

under acidic conditions (catalytic HCl in MeOH) to afford the target molecule 1 in 78% overall yield from 19. The physical and spectroscopic data of 1 were identical with those reported.¹⁷

2.3. Synthesis of β -dimorphecolic acid (2)

After completion of the synthesis of (-)-pinellic acid 1, we then further proceeded with the synthesis of β -dimorphecolic acid 2 from 1,3-envne product 21 (Scheme 4). Thus, coupling of 13 with trans-vinyl iodide 15 using Pd(PPh₃)₂Cl₂ and CuI in triethylamine furnished the 1.3envne product 21 in 86% yield. Reduction of 21 proceeded smoothly with the required (E)-geometry of the alkene using LAH in refluxing THF to afford 22 in good yield. The free hydroxy group of 22 was protected with TBDPSCl followed by deprotection of the PMB group with DDQ³⁰ to furnish compound 24 in good yield. Oxidation of the primary alcohol in 24 to the corresponding aldehyde using Swern conditions²⁴ and further oxidation using NaClO₂ in DMSO under buffer conditions²⁹ afforded the acid **25**. Finally, the TBDPS group was deprotected using TBAF to afford the target molecule 2 in 89% yield. The physical and spectroscopic data of 2 were identical with those reported.^{18,19,31}



Scheme 4. Reagents and conditions: (a) $Pd(PPh_3)_2Cl_2$, CuI, Et_3N , 8 h, 86%; (b) LAH, THF, 0 °C to rt, 83%; (c) TBDPSCl, imidazole, CH_2Cl_2 , 0 °C to rt, overnight, 96%; (d) DDQ, CH_2Cl_2/H_2O (9:1), 89%; (e) (i) ($COCl_2$, DMSO, Et_3N , CH_2Cl_2 , -78 °C to -60 °C; (ii) NaClO₂, DMSO, H_2O , NaH₂PO₄, rt, 1.5 h; (f) TBAF, THF, rt, overnight, 89%.

2.4. Synthesis of α-dimorphecolic acid

To achieve the synthesis of α -dimorphecolic acid, chiral propargylic alcohol 14 was converted into (E)-vinyl iodide through a vinyl stannane (Scheme 5). Thus, acetylene 14 was readily converted into (E)-vinyl stannane 26 by reaction with tri-n-butyltin hydride and AIBN in refluxing benzene in 99% yield. Tributyltin was then replaced with iodide by using I_2 in CH₂Cl₂³² to afford the corresponding iodo compound 27 in excellent yield. The Sonogashira coupling²² of 27 with commercially available 1-heptyne 28 was successfully carried out using Pd(PPh₃)₂Cl₂ and CuI in triethylamine to furnish the 1,3-envne product 29 in good yield. The partial hydrogenation of the triple bond in 29 proved to be challenging. Irrespective of whether substoichiometric quantities or several molar equivalents of quinoline were present, the mixture of 30 and over hydrogenated product was formed. The use of 1-octene as a co-solvent along with EtOAc in the presence of pyridine (EtOAc/pyridine/1-octene =10:1:1)³³ furnished the diene 30 as a single product. The subsequent deprotection of the *p*-methoxybenzyl group with DDQ³⁰ furnished the alcohol 31 in 94% yield. Oxidation of the resulting alcohol to the corresponding aldehyde using Swern conditions²⁴ and further oxidation using sodium chlorite in DMSO under buffer conditions²⁹ afforded the acid. Finally, the TBDPS group was deprotected using TBAF to afford the target molecule 3 in good yield. The physical and spectroscopic data of **3** were identical with those reported.¹⁸



Scheme 5. Reagents and conditions: (a) $(n-Bu)_3SnH$, AIBN, C_6H_6 , reflux, 4 h, 99%; (b) I_2 , CH_2CI_2 , 30 min, 96%; (c) Pd(PPh_3)_2CI_2, CuI, Et₃N, 1-hep-tyne (28), 6 h, 89%; (d) H₂, Lindlar's catalyst, EtOAc/pyridine/1-octene (10:1:1), 6 h, 95%; (e) DDQ, CH₂CI₂/H₂O (9:1), 94%; (f) (i) (COCI)₂, DMSO, Et₃N, CH₂CI₂, -78 °C to -60 °C, 95%; (ii) NaClO₂, DMSO, H₂O, NaH₂PO₄, rt, 1.5 h, 86%.

3. Conclusions

In conclusion, efficient total syntheses of (-)-pinellic acid 1, α - and β -dimorphecolic acid (2 and 3) with high enantioselectivity have been developed in which all the stereocentres were established by Sharpless asymmetric dihydroxylation. Notable features of this approach include Sonogashira coupling, reduction to establish the *trans*- and *cis*-olefin geometry. Further application of this methodology to the syntheses of related compounds for structure–activity relationship studies is currently underway in our laboratory.

4. Experimental

4.1. General methods

All reactions were carried out under argon or nitrogen in oven-dried glassware using standard gas-light syringes, cannulas and septa. Solvents and reagents were purified and dried by standard methods prior to use. Optical rotations were measured at room temperature. IR spectra were recorded on an FT-IR instrument. ¹H NMR spectra were recorded at 200 MHz, 300 MHz and 500 MHz and are reported in parts per million (δ) downfield relative to CDCl₃ as internal standard and ¹³C NMR spectra were recorded at 50 MHz, 75 MHz and 125 MHz and assigned in parts per million (δ) relative to CDCl₃. Column chromatography was performed on silica gel (100–200 and 230–400 mesh) using a mixture of petroleum ether and ethyl acetate as the eluent. Enantiomeric excess was determined using chiral HPLC.

4.1.1. 9-(4-Methoxybenzyloxy)nonan-1-ol (7). To a solution of 1,9-nonane diol 4 (8.0 g, 49.92 mmol) in dry DMF (200 mL) was added sodium hydride (50%, 2.64 g, 77.70 mmol) at 0 °C. The reaction mixture was then stirred at room temperature for 30 min after which it was again cooled to 0 °C. To this was added slowly *p*-methoxybenzyl bromide (10.04 g, 49.92 mmol) and tetra *n*-butylammonium iodide (1.84 g, 4.99 mmol) with further stirring for 1 h at the same temperature. The reaction mixture was quenched with addition of cold water at 0 °C. The two phases were separated and the aqueous phase was extracted with EtOAc ($3\times$ 100 mL). The combined organic layers were washed with water $(3 \times 100 \text{ mL})$, brine, dried (Na_2SO_4) and concentrated. The residual oil was purified by silica gel column chromatography using petroleum ether/EtOAc (8:2) as eluent to furnish the mono-PMB protected alcohol 7 (13.31 g, 95%) as a colourless oil. IR (CHCl₃): v=3400, 2937, 2863, 1612, 1513, 1248, 1174, 1097 cm⁻¹; ¹H NMR (200 MHz, CDCl₃): δ =1.27–1.42 (m, 10H), 1.52–1.64 (m, 4H), 3.49 (t, J=5.0 Hz, 2H), 3.62 (t, J= 5.0 Hz, 2H), 3.81 (s, 3H), 4.48 (s, 2H), 6.88 (d, J=10.0 Hz, 2H), 7.26 (d, J=10.0 Hz, 2H); ¹³C NMR (50 MHz, CDCl₃): $\delta = 13.9, 26.1, 27.8, 28.9, 29.1, 29.6, 55.1, 62.2, 69.9, 72.5,$ 113.8, 129.2, 130.4, 159.2. Anal. Calcd for C17H28O3 (280.40): C, 72.82; H, 10.06%. Found: C, 72.99; H, 9.87%.

4.1.2. (*E*)-Ethyl 11-(4-methoxybenzyloxy)undec-2-enoate

(8). To a solution of oxalyl chloride (4.75 g, 37.44 mmol) in dry CH₂Cl₂ (100 mL) at -78 °C was added dropwise dry DMSO (5.85 g, 5.3 mL, 74.89 mmol) in CH₂Cl₂ (20 mL). After 30 min, alcohol 7 (7.0 g, 24.96 mmol) in CH₂Cl₂ (20 mL) was added over 10 min giving a copious white precipitate. After stirring for 1 h at -78 °C the reaction mixture was brought to -60 °C and Et₃N (10.10 g, 13.9 mL, 99.85 mmol) was added slowly and stirred for 30 min allowing the reaction mixture to warm to room temperature. The reaction mixture was then diluted with water (150 mL) and CH₂Cl₂. The organic layer was separated and washed with water and brine, dried (Na₂SO₄) and passed through short pad of Celite. The filtrate was concentrated to give the aldehyde (9.86 g, 95%) as pale yellow oil, which was used as such for the next step without purification.

To a solution of (ethoxycarbonylmethylene)triphenyl-phosphorane (10.5 g, 30.17 mmol) in dry benzene (150 mL) was added a solution of the above aldehyde in dry benzene (100 mL). The reaction mixture was refluxed for 4 h. It was then concentrated and purified by silica gel column chromatography using petroleum ether/EtOAc (8.5:1.5) as eluent to afford the α , β -unsaturated olefin 8 (7.92 g, 91%) as a pale yellow liquid. IR (CHCl₃): ν =2956, 2858, 1724, 1654, 1038, 1300, 1204, 1100 cm⁻¹; ¹H NMR (200 MHz, CDCl₃): $\delta = 1.30 - 1.37$ (m, 11H), 1.48 - 1.68 (m, 4H), 2.28 (q, J=7.0 Hz, 2H), 3.48 (t, J=6.0 Hz, 2H), 3.85 (s, 3H), 4.25 (q, J=8.0 Hz, 2H), 4.48 (s, 2H), 5.90 (d, J=15.0 Hz, 1H), 6.91 (d, J=8.0 Hz, 2H), 6.98 (dt, J=15.0, 10.0 Hz, 1H), 7.30 (d, J=8.0 Hz, 2H); ¹³C NMR (50 MHz, CDCl₃): $\delta = 14.1, 26.0, 27.8, 28.9, 29.1, 29.6, 32.0, 54.9, 59.9, 69.9,$ 72.3, 113.5, 121.1, 128.9, 130.6, 149.1, 158.9, 166.5. Anal. Calcd for C₂₁H₃₂O₄ (348.48): C, 72.38; H, 9.26%. Found: C, 72.52; H, 9.11%.

4.1.3. (2R,3S)-Ethyl 11-(4-methoxybenzyloxy)-2,3-dihydroxyundecanoate (9). To a mixture of $K_3Fe(CN)_6$ (17.01 g, 51.65 mmol), K₂CO₃ (7.14 g, 51.65 mmol) and (DHQ)₂PHAL (134 mg, 1 mol %), in t-BuOH/H₂O (1:1, 175 mL) cooled at 0 °C was added OsO₄ (0.70 mL, 0.1 M solution in toluene, 0.4 mol %) followed by methanesulfonamide (1.64 g, 17.22 mmol). After being stirred for 5 min at 0 °C, the olefin 8 (6.0 g, 17.22 mmol) was added in one portion. The reaction mixture was stirred at 0 °C for 24 h and then quenched with solid sodium sulfite (25 g). The stirring was continued for an additional 45 min, and then the solution was extracted with EtOAc (3×50 mL). The combined organic extracts were washed with 10% KOH and brine, dried (Na₂SO₄) and concentrated. Silica gel column chromatography of the crude product using petroleum ether/EtOAc (3:2) as eluent gave the diol 9 (6.32 g, 96%) as a colourless syrupy liquid. The enantiomeric purity of the diol 9 was estimated to be >99% by chiral HPLC analysis (Chiralcel OD, petroleum ether/*i*-PrOH (96:4), 1 mL/min, 240 nm). $[\alpha]_D^{25}$ -6.7 (c 1.7, CHCl₃); IR (CHCl₃): v=3440, 2938, 2864, 1736, 1612, 1513, 1248, 1130, 1032 cm^{-1} ; ¹H NMR $(200 \text{ MHz}, \text{CDCl}_3)$: $\delta = 1.25 - 1.36 \text{ (m, 13H)}, 1.50 - 1.74 \text{ (m, 13H)}$ 4H), 2.51 (br s, 2H), 3.46 (t, J=6.6 Hz, 2H), 3.82 (s, 3H), 3.90 (d, J=6.6 Hz, 1H), 4.06–4.16 (m, 1H), 4.32 (q, J= 7.0 Hz, 2H), 4.46 (s, 2H), 6.93 (d, J=8.5 Hz, 2H), 7.31 (d, J=8.5 Hz, 2H); ¹³C NMR (50 MHz, CDCl₃): $\delta=14.0, 25.5,$ 26.0, 29.2, 29.3, 29.5, 33.5, 55.0, 61.6, 70.0, 72.3, 73.0, 73.1, 113.6, 128.9, 130.7, 158.9, 173.4. Anal. Calcd for C₂₁H₃₄O₆ (382.50): C, 65.94; H, 8.96%. Found: C, 66.09; H, 8.81%.

4.1.4. (4*S*,5*S*)-Ethyl 5-(8-(4-methoxybenzyloxy)octyl)-**2,2-dimethyl-1,3-dioxolane-4-carboxylate (10).** To a solution of the diol **9** (5.0 g, 13.07 mmol), *p*-TSA (100 mg) in CH₂Cl₂ (100 mL) was added 2,2-dimethoxypropane (2.0 g, 19.61 mmol) and the reaction mixture stirred overnight at room temperature. Solid NaHCO₃ (1 g) was added and the mixture again stirred for 30 min. The reaction mixture was filtered through a pad of neutral alumina and the filtrate concentrated. Silica gel column chromatography using petroleum ether/EtOAc (9:1) as eluent gave **10** (5.19 g, 94%) as a colourless liquid. $[\alpha]_D^{25} - 23.1$ (*c* 1.7, CHCl₃); IR (neat): ν =2864, 1736, 1612, 1513, 1248, 1130, 1032 cm⁻¹; ¹H NMR (200 MHz, CDCl₃): δ =1.29–1.40 (m, 13H), 1.46 (s, 3H), 1.49 (s, 3H), 1.56–1.63 (m, 2H), 1.66–1.80 (m, 2H), 3.45 (t, *J*=6.6 Hz, 2H), 3.82 (s, 3H), 4.11–4.14 (m, 2H), 4.25 (q, *J*=7.32 Hz, 2H), 4.45 (s, 2H), 6.91 (d, *J*=8.8 Hz, 2H), 7.30 (d, J=8.8 Hz, 2H); ¹³C NMR (50 MHz, CDCl₃): δ =13.8, 25.4, 25.6, 25.9, 29.2, 29.3, 29.6, 54.8, 60.9, 69.3, 72.3, 78.7, 110.5, 113.5, 128.8, 130.3, 158.9, 172.5. Anal. Calcd for C₂₄H₃₈O₆ (422.56): C, 68.22; H, 9.06%. Found: C, 68.01; H, 9.21%.

4.1.5. ((4S,5S)-5-(8-(4-Methoxybenzyloxy)octyl)-2,2-dimethyl-1,3-dioxolan-4-yl)methanol (11). To a solution of 10 (5.0 g, 11.83 mmol) in dry CH₂Cl₂ (80 mL) at 0 °C was added dropwise DIBAL-H (17.7 mL, 17.74 mmol, 1.0 M solution in toluene) through a syringe. The reaction mixture was allowed to warm to room temperature over 2 h, then re-cooled to 0 °C and treated with saturated solution of sodium/potassium tartrate. The solid material was filtered through a pad of Celite and concentrated. Silica gel column chromatography of the crude product using petroleum ether/ EtOAc (7:3) as eluent gave 11 (4.28 g, 95%) as a colourless oil. $[\alpha]_{D}^{25}$ -13.05 (c 1.6, CHCl₃); IR (CHCl₃): ν =3440, 2938, 2860, 1361, 1204, 1126, 1038 cm⁻¹; ¹H NMR (200 MHz, CDCl₃): $\delta = 1.28 - 1.40$ (m, 10H), 1.41 (s, 3H), 1.42 (s, 3H), 1.51-1.63 (m, 4H), 2.18 (s, 1H), 3.46-3.54 (m, 2H), 3.61 (dd, J=7.5, 4.4 Hz, 1H), 3.76 (m, 2H), 3.81 (s, 3H), 3.90 (dt, J=7.6, 4.0 Hz, 1H), 4.44 (s, 2H), 6.89 (d, J=8.7 Hz, 2H), 7.25 (d, J=8.7 Hz, 2H); ¹³C NMR (50 MHz, CDCl₃): $\delta =$ 25.4, 25.9, 26.9, 27.2, 29.3, 29.4, 29.7, 55.2, 62.1, 69.7, 72.4, 76.7, 81.6, 108.4, 113.7, 129.1, 130.5, 159.1. Anal. Calcd for C₂₂H₃₆O₅ (380.52): C, 69.44; H, 9.54%. Found: C, 69.62; H, 9.33%.

4.1.6. (4S,5R)-4-(8-(4-Methoxybenzyloxy)octyl)-5-(chloromethyl)-2,2-dimethyl-1,3-dioxolane (12). To a solution of alcohol 11 (5.0 g, 13.14 mmol) and Ph_3P (4.14 g, 15.77 mmol) in dry CH₂Cl₂ (50 mL) was added NCS (2.11 g, 15.77 mmol) at 0 °C under nitrogen. The reaction mixture was stirred at 0 °C for 1 h, then allowed to warm to room temperature, and stirred for 2 h. The mixture was diluted with 100 mL of hexane and passed through a pad of Celite under suction to remove the precipitate of Ph₃PO. The filtrate was concentrated, and resulting residue was dissolved in 100 mL of hexane and passed through a pad of Celite to remove the precipitate of Ph₃PO again. Evaporation of solvent gave 12 (4.67 g, 89%) as a colourless oil. $[\alpha]_D^{25}$ -5.6 $(c \ 0.5, \text{CHCl}_3)$; ¹H NMR (200 MHz, CDCl₃): $\delta = 1.29$ (br s, 10H), 1.40 (s, 3H), 1.41 (s, 3H), 1.52–1.61 (m, 4H), 3.42 (t, J=6.6 Hz, 2H), 3.57 (d, J=4.7 Hz, 2H), 3.79 (s, 3H), 3.84-3.88 (m, 2H), 4.42 (s, 2H), 6.88 (d, J=8.6 Hz, 2H), 7.23 (d, J=8.7 Hz, 2H); ¹³C NMR (50 MHz, CDCl₃): $\delta=25.8, 26.1,$ 27.0, 27.5, 29.3, 29.5, 29.7, 33.5, 44.4, 55.2, 70.2, 72.5, 79.4, 80.3, 109.1, 113.8, 129.1, 130.9, 159.1. Anal. Calcd for C₂₂H₃₅ClO₄ (398.96): C, 66.23; H, 8.84; Cl, 8.89%. Found: C, 66.45; H, 8.66; Cl, 8.92%.

4.1.7. (*S*)-11-(4-Methoxybenzyloxy)undec-1-yn-3-ol (13). To a solution of HMPA (11.43 g, 70.18 mmol) in dry THF (20 mL) was added *n*-BuLi (70.20 mL, 70.18 mmol, 1.0 M solution in hexane) at -42 °C under N₂. After 10 min, a solution of chloride 12 (4.0 g, 10.03 mmol) in THF (10 mL) was added dropwise over 5 min. The reaction mixture was warmed to room temperature and stirred for another 0.5 h. Saturated aqueous NH₄Cl solution was added to quench the reaction. The product was extracted with EtOAc (3× 30 mL), and combined organic layers were washed with brine, dried (Na₂SO₄) and concentrated. Silica gel column

chromatography of the crude product using petroleum ether/ EtOAc (7:3) as eluent gave **13** (2.50 g, 82%) as a colourless oil. $[\alpha]_{25}^{25}$ -1.75 (*c* 0.76, CHCl₃); ¹H NMR (200 MHz, CDCl₃): δ =1.33–1.51 (m, 10H), 1.59–1.79 (m, 4H), 1.86 (br s, 1H), 2.48 (d, *J*=2.0 Hz, 1H), 3.46 (t, *J*=6.6 Hz, 2H), 3.83 (s, 3H), 4.39 (dd, *J*=6.6, 1.8 Hz, 1H), 4.46 (s, 2H), 6.93 (d, *J*=8.2 Hz, 2H), 7.31 (d, *J*=8.5 Hz, 2H); ¹³C NMR (50 MHz, CDCl₃): δ =24.8, 25.9, 28.9, 29.1, 29.2, 29.4, 37.4, 55.0, 61.7, 69.9, 72.2, 72.4, 85.1, 113.5, 129.1, 130.4, 158.8. Anal. Calcd for C₁₉H₂₈O₃ (304.42): C, 74.96; H, 9.27%. Found: C, 74.81; H, 9.39%.

4.1.8. ((S)-11-(4-Methoxybenzyloxy)undec-1-yn-3-yloxy) (tert-butyl)diphenylsilane (14). A stirred solution of compound 13 (3.0 g, 9.85 mmol) and imidazole (1.0 g, 14.78 mmol) in dry CH₂Cl₂ (30 mL) was treated under argon with TBDPSCl (3.0 g, 10.84 mmol) at 0 °C and the reaction mixture was stirred overnight at the same temperature. The reaction mixture was quenched by addition of aqueous NH₄Cl (30 mL) and the mixture was extracted with CH_2Cl_2 (3×50 mL). The combined extracts were dried (Na₂SO₄) and concentrated. Silica gel column chromatography of the crude product using petroleum ether/EtOAc (9.5:0.5) as eluent gave the compound 14 (5.24 g, 98%) as a colourless syrupy liquid. $[\alpha]_{D}^{25}$ -22.3 (c 1.9, CHCl₃); IR $(CHCl_3)$: $\nu = 3440, 2938, 2864, 1736, 1612, 1513, 1248,$ 1130, 1032 cm⁻¹; ¹H NMR (200 MHz, CDCl₃): δ =1.10 (s, 9H), 1.24 (br s, 10H), 1.53–1.72 (m, 4H), 2.33 (d, J=2.0 Hz, 1H), 3.44 (t, J=6.6 Hz, 3H), 3.82 (s, 2H), 3.35 (ddd, J=6.1, 1.9 Hz, 1H), 4.45 (s, 2H), 6.92 (d, J=8.7 Hz, 2H), 7.31 (d, J=8.6 Hz, 2H), 7.37-7.45 (m, 6H), 7.64-7.79 (m, 4H); ¹³C NMR (50 MHz, CDCl₃): δ =19.2, 24.3, 24.5, 26.1, 26.5, 26.8, 28.9, 29.25, 29.28, 29.65, 37.2, 38.1, 55.0, 63.6, 69.2, 70.0, 72.4, 85.0, 113.6, 117.0, 127.3, 127.5, 129.1, 129.4, 129.56, 129.65, 129.7, 130.7, 133.9, 134.7, 135.7, 135.9, 159.0. Anal. Calcd for C₃₅H₄₆O₃Si (542.82): C, 77.44; H, 8.54%. Found: C, 77.51; H, 8.45%.

4.1.9. (*E*)-1-Iodohept-1-ene (15). Anhydrous $CrCl_2$ (22.09 g, 179.71 mmol) is suspended in THF (150 mL) under argon atmosphere. A solution of hexanaldehyde (3.0 g, 29.95 mmol) and iodoform (23.58 g, 59.90 mmol) in THF (30 mL) is added dropwise to the suspension at 0 °C. After stirring at 0 °C for 3 h, the reaction mixture is poured into water (100 mL) and extracted with ether (3×100 mL). The combined organic extracts were dried (Na₂SO₄) and concentrated. Silica gel column chromatography of the crude product using petroleum ether/EtOAc (9.5:0.5) as eluent gave 15 (5.90 g, 88%) with E/Z=95:5 selectivity as a pale yellow oil. ¹H NMR (200 MHz, CDCl₃): δ =0.92 (t, J=7.1 Hz, 3H), 1.26–1.43 (m, 6H), 1.62 (q, J=7.4 Hz, 1H), 2.00–2.18 (m, 1H), 6.01 (d, J=15.4 Hz, 1H), 6.44–6.59 (m, 1H); ¹³C NMR (50 MHz, CDCl₃): δ =14.1, 22.9, 28.5, 32.1, 34.4, 141.2, 82.1. Anal. Calcd for C₇H₁₃I (224.08): C, 37.52; H, 5.85; I, 56.63%. Found: C, 37.81; H, 5.72; I, 56.56%.

4.1.10. ((*S*,*E*)-1-(4-Methoxybenzyloxy)octadec-12-en-10yn-9-yloxy) (*tert*-butyl)diphenylsilane (16). To a stirred mixture of Pd(PPh₃)₂Cl₂ (181 mg, 0.257 mmol), CuI (147 mg, 0.77 mmol) in Et₃N (2 mL) were added solutions of (*E*)-1-iodohept-1-ene **15** (982 mg, 4.38 mmol) in Et₃N (2 mL) and acetylene **14** (1.2 g, 3.94 mmol) in Et₃N (2 mL) under argon. After 6 h, the reaction mixture was filtered through Celite and filtrate was concentrated. Silica gel column chromatography of the crude product using petroleum ether/EtOAc (9:1) as eluent gave 16 (1.36 g, 86%) as a pale yellow oil. $[\alpha]_{D}^{25}$ 3.5 (c 0.51, CHCl₃); IR (CHCl₃): *v*=2952, 2854, 1615, 1514, 1232, 1132, 1030 cm⁻¹; ¹H NMR (200 MHz, CDCl₃): δ =0.91 (t, J= 6.4 Hz, 3H), 1.09 (s, 9H), 1.22-1.43 (m, 16H), 1.56-1.72 (m, 4H), 2.06 (q, J=7.0 Hz, 2H), 3.44 (t, J=6.6 Hz, 2H), 3.81 (s, 3H), 4.44 (s, 2H), 4.47-4.54 (m, 1H), 5.38 (dd, J=15.8, 1.6 Hz, 1H), 5.76–5.97 (m, 1H), 6.91 (d, J=8.7 Hz, 2H), 7.30 (d, J=8.7 Hz, 2H), 7.36–7.43 (m, 6H), 7.68–7.79 (m, 4H); ¹³C NMR (50 MHz, CDCl₃): δ =14.0, 19.3, 22.4, 24.9, 26.1, 27.0, 28.4, 29.2, 29.4, 29.8, 31.2, 32.9, 38.4, 55.2, 64.4, 70.2, 72.5, 83.7, 89.2, 109.1, 113.7, 134.8, 135.9, 136.1, 144.0, 159.1. Anal. Calcd for C₄₂H₅₈O₃Si (638.99): C, 78.94; H, 9.15; Si, 4.40%. Found: C, 78.76; H, 9.33; Si, 4.58%.

4.1.11. (6S,7S,10S)-10-(tert-Butyldiphenylsilyloxy)-18-(4methoxybenzyloxy)octadec-8-yne-6,7-diol (17). To a mixture of K₃Fe(CN)₆ (1.55 g, 4.69 mmol), K₂CO₃ (649 mg, 4.69 mmol) and (DHQ)₂PHAL (12 mg, 1 mol %), in t-BuOH/H₂O (1:1, 20 mL) cooled at 0 °C was added OsO₄ (0.06 mL, 0.1 M solution in toluene, 0.4 mol %) followed by methanesulfonamide (149 mg, 1.56 mmol). After being stirred for 5 min at 0 °C, the olefin **16** (1.0 g, 1.56 mmol) was added in one portion. The reaction mixture was stirred at 0 °C for 24 h and then quenched with solid sodium sulfite (25 g). The stirring was continued for an additional 45 min, and then the solution was extracted with EtOAc $(3 \times 50 \text{ mL})$. The combined organic extracts were washed with brine, dried (Na₂SO₄) and concentrated. Silica gel column chromatography of the crude product using petroleum ether/EtOAc (3:2) as eluent gave the diol 17 (941 mg, 91%) as a colourless syrupy liquid. $[\alpha]_{D}^{25}$ +8.8 (c 0.9, CHCl₃); IR (CHCl₃): ν_{max} 3440, 2938, 2864, 1736, 1612, 1513, 1248, 1130, 1032 cm⁻¹; ¹H NMR (200 MHz, CDCl₃): δ 0.90 (t, J= 6.8 Hz, 3H), 1.07 (s, 9H), 1.27 (br s, 18H), 1.57-1.73 (m, 4H), 2.34 (br s, 2H), 3.29–3.34 (m, 2H), 3.81 (s, 3H), 3.90 (t, J=7.3 Hz, 2H), 4.23–4.34 (m, 1H), 4.45 (s, 2H), 6.91 (d, J=8.7 Hz, 2H), 7.30 (d, J=8.7 Hz, 2H), 7.35-7.49 (m, 6H), 7.69–7.78 (m, 4H); ¹³C NMR (50 MHz, CDCl₃): $\delta =$ 13.9, 19.1, 22.5, 24.8, 25.1, 26.1, 26.8, 29.0, 29.3, 29.4, 29.7, 31.7, 32.1, 38.2, 55.2, 63.7, 65.9, 70.1, 72.4, 74.5, 83.2, 87.7, 113.7, 127.3, 127.6, 129.2, 129.6, 129.8, 130.7, 133.4, 134.1, 135.8, 135.9, 159.1. Anal. Calcd for C₄₂H₆₀O₅Si (673.01): C, 74.95; H, 8.99; Si, 4.17%. Found: C, 75.10; H, 8.80; Si, 4.09%.

4.1.12. (9*S*,12*S*,13*S*,*E*)-9-(*tert*-Butyldiphenylsilyloxy)octadec-10-ene-1,12,13-triol (18). To the blue solution prepared by addition of lithium metal (334 mg, 48.18 mmol) to liquid NH₃ (5 mL) was added a solution of 17 (450 mg, 0.668 mmol) in dry THF (5 mL) at -78 °C. After the mixture was stirred for 1 h, the reaction was quenched by addition of 2.68 g (50.14 mmol) of NH₄Cl. After removal of NH₃ by a stream of N₂, the mixture was diluted with 50 mL of CHCl₃ and washed with water. The organic layer was dried (Na₂SO₄) and concentrated. Silica gel column chromatography of the crude product using petroleum ether/EtOAc (4:6) as eluent gave the triol 18 (333 mg, 90%) as a colourless syrupy liquid. [α]_D²⁵ 9.3 (*c* 0.42, CHCl₃); IR (CHCl₃): ν_{max} 3376, 1644, 1367, 1310, 1178, 1128, 1045, 980, 721 cm⁻¹; ¹H NMR (200 MHz, CDCl₃): δ =0.87 (t, *J*=6.3 Hz, 3H), 1.04 (s, 9H), 1.4 (br s, 18H), 1.51–1.72 (m, 4H), 2.35 (br s, 2H), 3.38–3.48 (m, 1H), 3.61 (t, *J*=6.6 Hz, 2H), 3.94 (t, *J*=5.9 Hz, 1H), 4.08 (q, *J*=6.6 Hz, 1H), 5.67 (dd, *J*=15.5, 5.9 Hz, 1H), 5.86 (dd, *J*=15.5, 5.6 Hz, 1H), 7.36–7.43 (m, 6H), 7.66–7.75 (m, 4H); ¹³C NMR (50 MHz, CDCl₃): δ =14.0, 22.6, 25.0, 26.1, 26.9, 28.1, 27.0, 28.1, 28.9, 29.1, 29.2, 31.8, 32.8, 32.9, 55.1, 64.1, 72.1, 74.6, 75.3, 127.5, 134.0, 134.8, 135.8, 136.1, 137.8. Anal. Calcd for C₃₄H₅₄O₄Si (554.88): C, 73.60; H, 9.81; Si, 5.06%. Found: C, 73.8; H, 9.64; Si, 5.33%.

4.1.13. (*S*,*E*)-9-(*tert*-Butyldiphenylsilyloxy)-11-((4*S*,5*S*)-2,2-dimethyl-5-pentyl-1,3-dioxolan-4-yl)undec-10-en-1-ol (19). Compound 19 was prepared following the procedure described for compound 10 in 92% yield as a colourless liquid. $[\alpha]_{D}^{25}$ 16.1 (*c* 0.57, CHCl₃); ¹H NMR (200 MHz, CDCl₃): δ =0.90 (t, *J*=6.3 Hz, 3H), 1.07 (s, 9H), 1.19–1.39 (br s, 18H), 1.41 (s, 3H), 1.45 (s, 3H), 1.54–1.73 (m, 4H), 2.31 (br s, 1H), 3.64 (t, *J*=6.6 Hz, 2H), 4.09 (q, *J*=6.6 Hz, 1H), 4.14–4.20 (m, 2H), 5.65 (dd, *J*=15.5, 7.1 Hz, 1H), 5.85 (dd, *J*=15.5, 5.7 Hz, 1H), 7.36–7.43 (m, 6H), 7.69–7.78 (m, 4H); ¹³C NMR (50 MHz, CDCl₃): δ =14.0, 19.3, 22.5, 24.9, 26.1, 26.9, 28.3, 29.2, 29.4, 29.7, 31.2, 31.9, 32.6, 64.1, 71.7, 80.1, 81.9, 108.6, 127.4, 134.0, 134.8, 135.8, 136.1, 137.8. Anal. Calcd for C₃₇H₅₈O₄Si (594.94): C, 74.70; H, 9.83; Si, 4.72%. Found: C, 75.01; H, 9.65; Si, 4.87%.

4.1.14. (*S*,*E*)-9-(*tert*-Butyldiphenylsilyloxy)-11-((4S,5S)-2,2-dimethyl-5-pentyl-1,3-dioxolan-4-yl)undec-10-enoic acid (20). To a solution of 19 (212 mg, 0.356 mmol) in EtOAc (5 mL) in 25 mL R.B. flask was added IBX (299 mg, 1.069 mmol) in one portion and the reaction mixture was refluxed for 5 h. The reaction mixture was filtered through a pad of Celite and the filtrate was concentrated to give the crude aldehyde, which was used in the next step without further purification.

A solution of 79% NaClO₂ (48 mg, 0.53 mmol) in 1.0 mL of water was added dropwise to a stirred solution of the above crude aldehyde (210 mg, 0.354 mmol) in 0.5 mL of DMSO and NaH₂PO₄ (84 mg, 0.71 mmol) in 1.0 mL of water in 5 min at room temperature. The mixture was left overnight at room temperature and then 5% aqueous solution of NaHCO₃ was added. The aqueous phase was extracted three times with CH_2Cl_2 and washed with brine, dried (Na₂SO₄) and concentrated to give the acid **20** (81%, 174 mg) as a yellowish liquid. $[\alpha]_D^{25}$ -9.3 (c 0.43, CHCl₃); ¹H NMR (200 MHz, CDCl₃): δ =0.89 (3H, t, J=6.3 Hz), 1.07 (s, 9H), 1.19-1.39 (br s, 16H), 1.41 (s, 3H), 1.45 (s, 3H), 1.50–1.62 (m, 4H), 2.26 (t, 2H, J=7.6 Hz), 3.41 (m, 1H), 4.12–4.16 (m, 2H), 5.66 (dd, 1H, J=15.6, 5.2 Hz), 5.71 (dd, 1H, J=15.6, 5.1 Hz), 7.36–7.43 (m, 6H), 7.69–7.78 (m, 4H); ¹³C NMR (50 MHz, CDCl₃): δ =177.6, 137.8, 136.1, 135.8, 134.8, 131.3, 134.0, 127.4, 108.6, 82.1, 80.3, 71.7, 38.8, 32.6, 31.9, 31.2, 29.7, 29.4, 29.2, 28.3, 26.9, 26.1, 24.9, 22.5, 19.3, 14.0. Anal. Calcd for C37H56O5Si (608.92): C, 72.98; H, 9.27%. Found: C, 73.10; H, 9.38%.

4.1.15. (9S,12S,13S)-(E)-9,12,13-Trihydroxy-10-octadecenoic acid [(-)-pinellic acid 1]. To a stirred solution of compound 20 (66 mg, 0.108 mmol) in MeOH was added a catalytic amount of concd HCl at room temperature and the

reaction mixture stirred overnight at the same temperature. The mixture was filtered through a Celite pad and washed with MeOH and concentrated. The crude product was redissolved in CH₂Cl₂ and washed with saturated aqueous NaHCO₃ and brine, dried (Na₂SO₄) and concentrated. Silica gel column chromatography of the crude product using CHCl₃/MeOH (10:1) furnished the target compound pinellic acid (-)-1 (28 mg, 78%) as a white solid. $R_{f}=0.24$ (silica gel, CHCl₃/MeOH/AcOH=10:1:0.1). Mp: 104-106 °C (lit.^{17a} mp: 104–106 °C); $[\alpha]_D^{25}$ –7.9 (c 0.30, MeOH) (lit.^{17a} $[\alpha]_D^{25}$ -8.1 (c 0.32, MeOH)); ¹H NMR (200 MHz, CDCl₃): $\delta = 0.89$ (t, J=6.3 Hz, 3H), 1.45–1.25 (m, 16H), 1.50–1.62 (m, 4H), 2.27 (t, J=7.6 Hz, 2H), 3.41 (m, 1H), 3.91 (dd, J=5.5, 5.0 Hz, 1H), 4.05 (m, 1H), 5.65 (dd, J=15.6, 5.2 Hz, 1H), 5.72 (dd, J=15.6, 5.1 Hz, 1H); ¹³C NMR (50 MHz, CDCl₃): $\delta = 177.6$, 136.6, 131.2, 76.4, 75.8, 73.0, 38.3, 35.1, 33.6, 33.2, 30.5, 30.4, 30.1, 26.6, 26.5, 26.1, 23.7, 13.9.

4.1.16. (S,E)-1-(4-Methoxybenzyloxy)octadec-12-en-10yn-9-ol (21). To a stirred mixture of Pd(PPh₃)₂Cl₂ (276 mg, 0.39 mmol), CuI (225 mg, 1.18 mmol) in Et₃N (2 mL) were added solutions of (E)-1-iodohept-1-ene 15 (1.50 g, 6.70 mmol) in Et_3N (2 mL) and acetylene 13 (1.2 g, 3.94 mmol) in Et₃N (2 mL) under argon. After 8 h, the reaction mixture was filtered through Celite and the filtrate was concentrated. Silica gel column chromatography of the crude product using petroleum ether/EtOAc (9:1) as eluent gave 21 (1.36 g, 86%) as a pale yellow oil. $[\alpha]_{D}^{25}$ 1.75 (c 0.8, CHCl₃); IR (CHCl₃): v=3440, 2938, 2864, 1736, 1612, 1513, 1248, 1130, 1032 cm⁻¹; ¹H NMR (200 MHz, CDCl₃): δ =0.89 (t, J=6.7 Hz, 3H), 1.20–1.52 (m, 16H), 1.53-1.75 (m, 4H), 2.01-2.20 (m, 2H), 3.44 (t, J=6.6 Hz, 2H), 3.81 (s, 3H), 4.12-4.26 (m, 1H), 4.44 (s, 2H), 5.52 (dd, J=15.9, 1.64 Hz, 1H), 6.07-6.26 (m, 1H), 6.90 (d, J=8.7 Hz, 2H), 7.29 (d, J=8.7 Hz, 2H); ¹³C NMR (50 MHz, CDCl₃): δ =14.0, 22.5, 25.4, 25.6, 25.7, 28.3, 29.2, 29.4, 29.7, 31.2, 32.9, 38.4, 55.2, 64.4, 70.2, 72.5, 83.7, 89.2, 109.1, 113.7, 129.1, 130.9, 144.4, 159.1. Anal. Calcd for C₂₆H₄₀O₃ (400.59): C, 77.95; H, 10.06%. Found: C, 78.21; H, 10.11%.

4.1.17. (S,10E,12E)-1-(4-Methoxybenzyloxy)-octadeca-10,12-dien-9-ol (22). To a suspension of lithium aluminium hydride (61 mg, 1.62 mmol) in THF (40 mL) was added a solution of **21** (650 mg, 1.62 mmol) in THF (40 mL) at 0 °C. The mixture was warmed to room temperature and was stirred for 10 min. The mixture was refluxed for 1 h and then cooled to 0 °C. The reaction was quenched with saturated aqueous potassium sodium tartrate solution. After the suspension was stirred vigorously, the aqueous layer was extracted with ether. The ethereal extract was washed with brine, dried (Na_2SO_4) and concentrated. Silica gel column chromatography of the crude product using petroleum ether/EtOAc (9:1) as eluent gave 22 (542 mg, 83%) as a pale yellow oil. $[\alpha]_D^{25}$ 3.1 (c 0.4, CHCl₃); IR (CHCl₃): *v*=3400, 2938, 2863, 1612, 1513, 1457, 1216, 1054, 990; ¹H NMR (200 MHz, CDCl₃): δ =0.89 (t, J=6.7 Hz, 3H), 1.20-1.50 (m, 16H), 1.53-1.76 (m, 4H), 2.08 (q, J=6.8 Hz, 2H), 3.44 (t, J=6.6 Hz, 2H), 3.80 (s, 3H), 4.05–4.24 (m, 1H), 4.45 (s, 2H), 5.53 (dd, J=15.9, 6.5 Hz, 1H), 5.71 (m, 1H), 6.01 (dd, J=15.1, 6.6 Hz, 1H), 6.15 (dd, J=15.2, 10.5 Hz, 1H), 6.90 (d, *J*=8.7 Hz, 2H), 7.29 (d, *J*=8.7 Hz, 2H); ¹³C NMR (50 MHz, CDCl₃): δ =13.8, 22.5, 25.4, 25.7, 28.9,

29.3, 29.4, 29.5, 31.4, 32.6, 32.8, 37.3, 55.0, 72.9, 73.1, 113.5, 129.1, 129.5, 130.1, 131.0, 133.6, 135.8, 158.8. Anal. Calcd for $C_{26}H_{42}O_3$ (400.59); C, 77.56; H, 10.51%. Found: C, 77.62; H, 10.33%.

4.1.18. ((S,10E,12E)-1-(4-Methoxybenzyloxy)-octadeca-10,12-dien-9-yloxy) (tert-butyl)diphenylsilane (23). Compound 23 was prepared following the procedure described for compound 14 in 96% yield as a colourless liquid. $[\alpha]_D^{25}$ +19.5 (c 0.7, CHCl₃); IR (CHCl₃): ν =2938, 2864, 1736, 1612, 1513, 1248, 1130, 1032 cm⁻¹; ¹H NMR (200 MHz, CDCl₃): δ =0.89 (t, J=6.7 Hz, 3H), 1.04 (s, 9H), 1.19–1.50 (m, 16H), 1.53–1.81 (m, 4H), 2.07 (q, J=7.0 Hz, 2H), 3.45 (t, J=6.6 Hz, 2H), 3.81 (s, 3H), 4.10 (q, J=6.8 Hz, 1H), 4.44 (s, 2H), 5.56 (dd, J=15.2, 6.6 Hz, 1H), 5.70 (m, 1H), 6.01 (dd, J=15.1, 6.8 Hz, 1H), 6.16 (dd, J=15.2, 10.5 Hz, 1H), 6.89 (d, J=8.7 Hz, 2H), 7.29 (d, J=8.7 Hz, 2H), 7.36-7.46 (m, 6H), 7.64–7.79 (m, 4H); ¹³C NMR (50 MHz, CDCl₃): δ =13.9, 19.2, 22.5, 25.7, 25.8, 26.9, 28.9, 29.2, 29.5, 29.7, 31.4, 32.5, 32.7, 37.9, 54.9, 64.3, 70.3, 72.5, 113.8, 127.5, 127.6, 127.7, 129.1, 129.4, 129.5, 129.6, 130.1, 131.1, 133.4, 134.0, 135.7, 135.9, 159.0. Anal. Calcd for C₄₂H₆₀O₃Si (641.01): C, 78.70; H, 9.43; Si, 4.38%. Found: C, 78.88; H, 9.29; Si, 4.51%.

4.1.19. (S.10E.12E)-9-(tert-Butyldiphenvlsilvloxy)-octadeca-10,12-dien-1-ol (24). To a stirring solution of PMB ether 22 (380 mg, 0.59 mmol) in CH₂Cl₂/H₂O (10 mL/ 0.5 mL, 20:1) was added DDQ (161 mg, 0.71 mmol). The resulting mixture was stirred for 45 min at room temperature. The mixture was poured into saturated aqueous NaHCO₃ and further diluted with CH₂Cl₂. The layers were separated and the aqueous layer was extracted with CH_2Cl_2 (2×10 mL). The combined organic layers were dried (Na₂SO₄), filtered and concentrated. The residue was then filtered through a pad of Celite and washed with 50% EtOAc/hexane (20 mL). The solvents were removed under reduced pressure to give the crude product mixture as yellow oil. Silica gel column chromatography of the crude product using petroleum ether/EtOAc (7:3) as eluent gave 24 (275 mg, 89%) as a colourless oil. $[\alpha]_{D}^{25}$ +6.28 (c 0.9, CHCl₃); IR (CHCl₃): ν =3400, 2928, 2855, 1659, 1589, 1464, 1427, 1389, 1216, 1112, 986 cm⁻¹; ¹H NMR (200 MHz, CDCl₃): δ =0.89 (t, J=6.7 Hz, 3H), 1.04 (s, 9H), 1.19–1.50 (m, 16H), 1.53–1.81 (m, 4H), 2.08 (q, J=7.0 Hz, 2H), 3.64 (t, J=6.6 Hz, 2H), 4.12 (q, J=6.6 Hz, 1H), 5.56 (dd, J=15.2, 6.6 Hz, 1H), 5.70 (dt, J=15.2, 7.1 Hz, 1H), 6.02 (dd, J=15.1, 10.5 Hz, 1H), 6.16 (dd, J=15.2, 10.4 Hz, 1H), 7.36-7.45 (m, 6H), 7.66-7.78 (m, 4H); ¹³C NMR (50 MHz, CDCl₃): δ =14.1, 19.2, 22.5, 25.6, 25.8, 26.9, 28.8, 29.5, 29.7, 31.5, 32.6, 32.7, 37.4, 64.1, 73.1, 127.6, 129.4, 129.5, 131.1, 133.6, 134.3, 135.7. Anal. Calcd for C₃₄H₅₂O₂Si (520.86): C, 78.40; H, 10.06; Si, 5.39%. Found: C, 78.61; H, 9.87; Si, 5.52%.

4.1.20. (9S,10E,12Z)-9-Hydroxy-10,12-octadecadienoic acid [β -dimorphecolic acid] (2). To a solution of oxalyl chloride (76 mg, 0.05 mL, 0.60 mmol) in dry CH₂Cl₂ (20 mL) at -78 °C was added dropwise dry DMSO (94 mg, 0.095 mL, 1.90 mmol) in CH₂Cl₂ (5 mL). After 30 min, alcohol 24 (210 mg, 0.40 mmol) in CH₂Cl₂ (5 mL) was added over 10 min giving a copious white precipitate. After stirring for 1 h at -78 °C the reaction mixture was brought to -60 °C and Et₃N (0.163 g, 0.22 mL,

1.61 mmol) was added slowly and stirred for 30 min allowing the reaction mixture to warm to room temperature. The reaction mixture was poured into water (50 mL) and the organic layer was separated. The aqueous layer was extracted with CH_2Cl_2 (2×25 mL) and combined organic layers were washed with water (3×20 mL), brine (20 mL), dried (Na₂SO₄) and passed through short pad of silica gel. The filtrate was concentrated to give the aldehyde (302 mg) as pale yellow syrup, which was used as such for the next step without further purification.

A solution of 79% NaClO₂ (68 mg, 0. 60 mmol) in 1.0 mL of water was added dropwise to a stirred solution of crude aldehyde (209 mg, 0.40 mmol) in 0.5 mL of DMSO and NaH₂PO₄ (97 mg, 0.81 mmol) in 1.0 mL of water in 5 min at room temperature. The mixture was left overnight at room temperature and then 5% aqueous solution of NaHCO₃ was added. The aqueous phase was extracted three times with CH₂Cl₂ and washed with brine, dried (Na₂SO₄) and concentrated to give the crude product **25**, which was used as such for the next step without further purification.

The above crude acid 25 (215 mg, 0.40 mmol) was dissolved in THF (5 mL), followed by the dropwise addition of TBAF (0.80 mL, 1 M solution in THF, 0.80 mmol). The reaction mixture was stirred at room temperature overnight and then quenched by addition of water, and the aqueous layer was extracted with EtOAc (3×30 mL) and combined EtOAc extracts were washed with brine, dried (Na₂SO₄) and concentrated. Silica gel column chromatography of the crude product using petroleum ether/EtOAc (9:1) as eluent gave β -dimorphecolic acid (2) (82 mg, 69% yield over three steps) as a cream-coloured solid. Mp 38-40 °C (lit.¹⁹ 39-40 °C); $[\alpha]_D^{25}$ +15.1 (c 0.8, in MeOH) [lit.¹⁹ $[\alpha]_D^{25}$ +15.4 (c 5.0, in MeOH)]; IR (KBr): v=3422, 2924, 2869, 1712, 1459, 1321, 1211, 986 cm⁻¹; ¹H NMR (200 MHz, CD₃OD): δ =0.89 (t, J=6.8 Hz, 3H), 1.19–1.51 (m, 16H), 1.53–1.81 (m, 4H), 2.06 (q, J=7.0 Hz, 2H), 2.24 (t, J=7.5 Hz, 2H), 4.02 (q, J=6.6 Hz, 1H), 5.51 (dd, J=15.1, 6.5 Hz, 1H), 5.66 (m, 1H), 6.02 (dd, J=15.1, 10.5 Hz, 1H), 6.14 (dd, J=15.1, 10.5 Hz, 1H); ¹³C NMR (50 MHz, CDCl₃): $\delta = 14.0, 22.5, 24.9, 25.3, 28.9, 29.1, 29.3, 29.7, 31.4, 32.6,$ 34.1, 37.2, 72.8, 129.4, 131.0, 133.5, 135.6, 178.5.

4.1.21. ((S,E)-11-(4-Methoxybenzyloxy)-1-(tributyl-stannyl)undec-1-en-3-vloxy) (tert-butyl)diphenylsilane (26). To a stirred solution of 14 (1.80 g, 3.32 mmol) in benzene (30 mL) were added *n*-Bu₃SnH (1.1 mL, 3.98 mmol) and AIBN (catalytic) at room temperature under N2. The reaction mixture was gently refluxed with stirring for 4 h. The solvent was removed and the residue was purified by silica gel column chromatography using petroleum ether/EtOAc (9:1) as eluent to give **26** (2.74 g, 99%) as yellowish oil. $[\alpha]_D^{25} - 9.8$ (c 0.64, CHCl₃); IR (CHCl₃): v=3004, 2957, 2928, 2853, 1612, 1513, 1464, 1378, 1247, 1173, 1036 cm⁻¹; ¹H NMR (200 MHz, CDCl₃): δ=0.88-0.93 (m, 9H), 1.10 (s, 9H), 1.24 (br s, 10H), 1.26-1.31 (m, 10H), 1.42-1.74 (m, 12H), 3.44 (t, J=6.6 Hz, 2H), 3.82 (s, 3H), 4.12 (q, J=6.6 Hz, 1H), 4.45 (s, 2H), 5.95 (dd, J=19.0, 5.1 Hz, 1H), 6.36 (d, J=19.0 Hz, 1H), 6.92 (d, J=8.7 Hz, 2H), 7.31 (d, J=8.6 Hz, 2H), 7.37-7.45 (m, 6H), 7.64-7.79 (m, 4H); ¹³C NMR $(50 \text{ MHz}, \text{CDCl}_3)$: $\delta = 8.9, 9.4, 10.2, 10.6, 11.2, 13.5, 19.3,$ 24.2, 24.5, 26.1, 26.4, 26.9, 28.9, 29.3, 29.3, 29.7, 37.2,

38.1, 55.2, 61.8, 69.9, 72.3, 113.6, 117.0, 127.2, 127.3, 127.4, 129.2, 129.5, 129.56, 129.65, 129.7, 130.8, 133.9, 134.5, 134.9, 135.7, 135.9, 144.9, 159.1. Anal. Calcd for $C_{47}H_{74}O_3SiSn$ (833.88): C, 67.70; H, 8.94; Si, 3.37%. Found: C, 67.59; H, 9.21; Si, 3.62%.

4.1.22. ((S,E)-11-(4-Methoxybenzyloxy)-1-iodoundec-1en-3-yloxy) (tert-butyl)diphenylsilane (27). To a cooled $(0 \circ C)$, stirred solution of **26** (1.60 g, 1.92 mmol) in CH₂Cl₂ (40 mL) was added iodine (973 mg, 3.84 mmol). After 30 min at 0 °C, the reaction mixture was diluted with CH₂Cl₂, washed with saturated Na₂S₂O₃ and 10% KF solutions, and brine. The organic layer was dried (Na_2SO_4) , filtered and concentrated. Silica gel column chromatography of the crude product using petroleum ether/EtOAc (9.5:0.5) as eluent gave 27 (1.24 g, 96%) as a yellowish oil. $[\alpha]_D^{25}$ -6.1 (c 0.48, CHCl₃); IR (CHCl₃): v=2948, 2933, 2861, 1612, 1515, 1465, 1372, 1174, 1092, 948 cm⁻¹; ¹H NMR $(200 \text{ MHz}, \text{CDCl}_3)$: $\delta = 1.10 \text{ (s, 9H)}, 1.24 \text{ (br s, 10H)}, 1.53 \text{--}$ 1.72 (m, 4H), 3.47 (t, J=6.1 Hz, 2H), 3.92 (s, 3H), 4.12 (q, J=6.6 Hz, 1H), 4.45 (s, 2H), 6.29 (m, 2H), 6.53 (d, J=8.7 Hz, 2H), 7.31 (d, J=8.6 Hz, 2H), 7.37–7.45 (m, 6H), 7.64–7.79 (m, 4H); ¹³C NMR (50 MHz, CDCl₃): δ =19.2, 24.3, 24.5, 26.1, 26.5, 26.8, 28.9, 29.25, 29.28, 29.65, 37.2, 38.1, 55.0, 62.8, 70.1, 72.2, 114.0, 117.0, 127.2, 127.3, 127.4, 128.6, 129.5, 129.6, 129.7, 130.8, 133.9, 134.5, 134.9, 135.7, 135.9, 147.6, 161.4. Anal. Calcd for C₃₅H₄₇IO₃Si (670.74): C, 62.67; H, 7.06; I, 18.92%. Found: C, 62.89; H, 6.96; I, 19.24%.

4.1.23. ((S,E)-1-(4-Methoxybenzyloxy)octadec-10-en-12vn-9-vloxv) (tert-butvl)diphenvlsilane (29). Compound 29 was prepared following the procedure described for compound 21 in 89% yield as a pale yellow oil. $[\alpha]_D^{25}$ -15.4 (c 0.4, CHCl₃); IR (CHCl₃): $\nu = 3004$, 2957, 2928, 2853, 1612, 1513, 1464, 1378, 1247, 1173, 1036 cm⁻¹; ¹H NMR (200 MHz, CDCl₃): δ=0.89 (t, J=6.7 Hz, 3H), 1.04 (s, 9H), 1.20-1.52 (m, 16H), 1.53-1.81 (m, 4H), 2.08 (q, J=6.8 Hz, 2H), 3.49 (t, J=6.6 Hz, 2H), 3.81 (s, 3H), 4.14-4.28 (m, 1H), 4.44 (s, 2H), 5.72 (dd, J=15.9, 1.64 Hz, 1H), 6.03-6.24 (m, 1H), 6.91 (d, J=8.7 Hz, 2H), 7.29 (d, J=8.7 Hz, 2H), 7.36–7.45 (m, 6H), 7.66–7.78 (m, 4H); ¹³C NMR (50 MHz, CDCl₃): δ=14.0, 19.2, 22.6, 25.7, 25.8, 26.9, 28.9, 29.2, 29.5, 29.7, 31.4, 32.5, 32.7, 37.9, 55.1, 64.6, 71.3, 72.5, 81.8, 82.4, 113.8, 127.5, 127.6, 127.7, 129.0, 129.4, 129.6, 130.5, 132.5, 133.8, 134.0, 135.7, 135.9, 159.1. Anal. Calcd for C₄₂H₅₈O₃Si (638.99): C, 78.94; H, 9.15; Si, 4.40%. Found: C, 79.22; H, 9.01; Si, 4.67%.

4.1.24. ((*S*,10*E*,12*Z*)-1-(4-Methoxybenzyloxy)-octadeca-10,12-dien-9-yloxy) (*tert*-butyl)diphenylsilane (30). To a solution of **29** (620 mg, 0.97 mmol) in 10 mL of ethyl acetate/pyridine/1-octene (10:1:1) was added Lindlar's catalyst (20 mg). The reaction mixture was stirred for 6 h under a balloon of H₂ at room temperature and filtered through a Celite pad. The filtrate was concentrated and the residue was purified by silica gel column chromatography using petroleum ether/EtOAc (9:1) as eluent to give **30** (590 mg, 95%) as a pale yellow oil. $[\alpha]_D^{25}$ +12.59 (*c* 0.43, CHCl₃); IR (CHCl₃): ν =3600, 1715, 985, 950 cm⁻¹; ¹H NMR (200 MHz, CDCl₃): δ =0.90 (t, *J*=6.8 Hz, 3H), 1.04 (s, 9H), 1.19–1.51 (m, 16H), 1.54–1.80 (m, 4H), 2.09 (q, *J*= 7.1 Hz, 2H), 3.45 (t, *J*=6.6 Hz, 2H), 3.81 (s, 3H), 4.12 (q, J=6.9 Hz, 1H), 4.41 (s, 2H), 5.49 (dt, J=11.0, 7.1 Hz, 1H), 5.64 (dd, J=15.0, 7.0 Hz, 1H), 6.01 (dd, J=11.0, 11.0 Hz, 1H), 6.51 (dd, J=15.1, 11.0 Hz, 1H), 6.90 (d, J=8.7 Hz, 2H), 7.30 (d, J=8.7 Hz, 2H), 7.36–7.48 (m, 6H), 7.66–7.79 (m, 4H); ¹³C NMR (50 MHz, CDCl₃): δ =13.9, 19.2, 22.5, 25.7, 25.8, 26.9, 28.9, 29.2, 29.5, 29.7, 31.4, 32.5, 32.7, 37.9, 54.9, 64.9, 70.5, 72.3, 113.8, 113.8, 126.1, 127.5, 127.6, 127.9, 129.1, 129.4, 129.5, 129.6, 130.1, 131.1, 133.4, 134.0, 135.7, 135.9, 159.0. Anal. Calcd for C₄₂H₆₀O₃Si (641.01): C, 78.70; H, 9.43; Si, 4.38%. Found: C, 78.84; H, 9.28; Si, 4.41%.

4.1.25. (S.10E.12Z)-9-(tert-Butyldiphenylsilyloxy)-octadeca-10,12-dien-1-ol (31). Compound 31 was prepared following the procedure described for compound 25 in 94% yield as a colourless liquid. $[\alpha]_D^{25}$ +10.1 (c 0.39, CHCl₃); IR $(CHCl_3)$: $\nu = 3400, 2928, 2855, 1659, 1589, 1464, 1427,$ 1389, 1216, 1112, 986, 950, 985 cm⁻¹; ¹H NMR (200 MHz. CDCl₃): δ=0.89 (t, J=6.7 Hz, 3H), 1.05 (s, 9H), 1.19–1.52 (m, 16H), 1.53–1.79 (m, 4H), 2.21 (q, J=7.0 Hz, 2H), 3.68 (t, J=6.6 Hz, 2H), 4.07 (q, J=7.1 Hz, 1H), 5.56 (dd, J=15.2, 6.6 Hz, 1H), 5.70 (m, 2H), 6.02 (dd, J=15.1, 10.5 Hz, 1H), 7.36–7.45 (m, 6H), 7.66–7.78 (m, 4H); ¹³C NMR (50 MHz, CDCl₃): δ =14.1, 19.2, 22.9, 25.7, 25.8, 26.9, 28.8, 29.5, 29.7, 31.4, 32.6, 32.7, 37.8, 64.4, 73.1, 126.3, 127.6, 127.9, 129.4, 129.5, 133.2, 134.1, 135.7. Anal. Calcd for C₃₄H₅₂O₂Si (520.86): C, 78.40; H, 10.06; 6.14; Si, 5.39%. Found: C, 78.53; H, 10.28; Si, 5.58%.

4.1.26. α-Dimorphecolic acid (3). Compound **3** was prepared following the procedure described for compound **2** in 89% yield as a yellow syrup. $[\alpha]_D^{25} +11.4$ (*c* 0.4, in MeOH); IR (KBr): ν =3600, 1715, 950, 985 cm⁻¹; ¹H NMR (200 MHz, CD₃OD): δ =0.90 (t, *J*=6.7 Hz, 3H), 1.19–1.50 (m, 16H), 1.52–1.79 (m, 4H), 2.20 (m, 2H), 2.36 (t, *J*=7.0 Hz, 2H), 4.17 (q, *J*=7.0 Hz, 1H), 5.47 (dt, *J*=11.0, 7.1 Hz, 1H), 5.66 (dd, *J*=15.0, 7.0 Hz, 1H), 5.98 (dd, *J*=11.0, 11.0 Hz, 1H), 6.51 (dd, *J*=15.1, 11.0 Hz, 1H); ¹³C NMR (50 MHz, CD₃OD): δ =14.1, 22.5, 25.6, 25.8, 26.9, 28.8, 29.5, 29.7, 31.5, 32.6, 32.7, 37.4, 73.0, 126.2, 127.9, 133.2, 135.9, 178.6.

Acknowledgements

S.V.N. and P.G. thank CSIR and UGC New Delhi for financial assistance. We are grateful to Dr. M. K. Gurjar for his support and encouragement. The financial support from DST, New Delhi (Grant No. SR/S1/OC-40/2003) is gratefully acknowledged. This is NCL communication no. 6702.

References and notes

- Murphy, B. R.; Webster, R. G. Orthomyxoviruses. In *Virology*, 2nd ed.; Fields, B. N., Knipe, D. M., Eds.; Raven: New York, NY, 1990; pp 1091–1152.
- (a) Miyamoto, T. Asian Med. J. 1992, 35, 30–36; (b) Nagai, T.; Yamada, H. Int. J. Immunopharmacol. 1994, 16, 605–613;
 (c) Nagai, T.; Urata, M.; Yamada, H. Immunopharmacol. Immunotoxicol. 1996, 18, 193–208.
- (a) Kato, T.; Yamaguchi, Y.; Ohnuma, S.; Uyehara, T.; Namai, T.; Kodama, M.; Shiobara, Y. *Chem. Lett.* **1986**, 577–580;

(b) Colin, D. F.; Wiliam, S. P. *Biochim. Biophys. Acta* **1983**, 754, 57–71; (c) Hanberg, M. *Lipids* **1991**, 26, 407–415; (d) Harada, N.; Iwabuchi, J.; Yokota, Y.; Uda, H.; Nakanishi, K. *J. Am. Chem. Soc.* **1981**, *103*, 5590–5591.

- Shirahata, T.; Sunazuka, T.; Yoshida, K.; Yamamoto, D.; Harigaya, Y.; Kuwajima, I.; Nagai, T.; Kiyohara, H.; Yamada, H.; Omura, S. *Tetrahedron* 2006, *62*, 9483–9496.
- Smith, C. R., Jr.; Wilson, T. L.; Melvin, E. H.; Wolff, I. A. J. Am. Chem. Soc. 1960, 82, 1417–1421.
- Hopkins, C. Y.; Chilsholm, M. J. Can. J. Chem. 1965, 43, 3160– 3164.
- Henry, D. Y.; Gueritte-Voegelein, F.; Insel, P. A.; Ferry, N.; Bouguet, J.; Potier, P.; Sevenet, T.; Hanoune, J. *Eur. J. Biochem.* 1987, 170, 389–394.
- 8. Blondin, G. A. Ann. N.Y. Acad. Sci. 1975, 264, 98-111.
- Takashi, M.; Hiroshi, S.; Masao, C.; Shunji, S.; Naoko, K.; Kiyoshi, F.; Hiroshi, M. JP 62-164620, 1987; *Chem. Abstr.* 1988, 108, 26976.
- 10. Kraus, R.; Spiteller, G.; Bartsch, W. *Liebigs Ann. Chem.* **1991**, 335–339.
- Aisen, P. S.; Haines, K. A.; Given, W.; Abramson, S. B.; Pras, M.; Serhan, C.; Hamberg, M.; Samuelsson, B.; Weissmann, G. *Proc. Natl. Acad. Sci. U.S.A.* **1985**, *82*, 1232–1236.
- Chisholm, M. J.; Hopkins, C. Y. Can. J. Chem. 1960, 38, 2500– 2507.
- Powell, R. G.; Smith, C. R., Jr.; Wolff, I. A. J. Org. Chem. 1967, 32, 1442–1446.
- Kato, T.; Yamaguchi, Y.; Hirano, T.; Yokoyama, T.; Uyehara, T.; Namai, T.; Yamanaka, S.; Harada, N. Chem. Lett. 1984, 409–412.
- Duffault, J. M.; Einhorn, J.; Alexakis, A. *Tetrahedron Lett.* 1995, 36, 2765–2768 and references therein.
- 16. Mann, J. Secondary Metabolism; Clarendon: Oxford, 1987.
- (a) Sunazuka, T.; Shirahata, T.; Yoshida, K.; Yamamoto, D.; Harigaya, Y.; Nagai, T.; Kiyohara, H.; Yamada, H.; Kuwajimad, I.; Ōmura, S. *Tetrahedron Lett.* **2002**, *48*, 1265– 1268; (b) Shirahata, T.; Sunazuka, T.; Yoshida, K.; Yamamoto, D.; Harigaya, Y.; Nagai, T.; Kiyohara, H.; Yamada, H.; Kuwajimad, I.; Ōmura, S. *Bioorg. Med. Chem. Lett.* **2003**, *13*, 937–941; (c) Sabitha, G.; Reddy, E. V.; Bhikshapathi, M.; Yadav, J. S. *Tetrahedron Lett.* **2007**, *43*, 313–315.
- (a) Rama Rao, A. V.; Reddy, E. R.; Sharma, G. V. M.; Yadagiri,
 P.; Yadav, J. S. *Tetrahedron Lett.* **1985**, *26*, 465–468; (b)
 Tsuboi, S.; Maeda, S.; Takeda, A. *Bull. Chem. Soc. Jpn.* **1986**, *59*, 2050–2052.
- (a) Shimazaki, T.; Kobayashi, Y.; Sato, F. Chem. Lett. 1988, 10, 1785–1788; (b) Ley, S. V.; Meek, G. J. Chem. Soc., Perkin Trans. 1 1997, 1125–1133.
- (a) Kandula, S. V.; Kumar, P. *Tetrahedron Lett.* 2003, 44, 6149–6151; (b) Gupta, P.; Naidu, S. V.; Kumar, P. *Tetrahedron Lett.* 2004, 45, 849–851; (c) Naidu, S. V.; Gupta, P.; Kumar, P.

Tetrahedron Lett. **2005**, *46*, 2129–2131; (d) Kumar, P.; Naidu, S. V.; Gupta, P. J. Org. Chem. **2005**, *70*, 2843–2846; (e) Kumar, P.; Naidu, S. V. J. Org. Chem. **2005**, *70*, 4207– 4210; (f) Gupta, P.; Naidu, S. V.; Kumar, P. *Tetrahedron Lett.* **2005**, *46*, 6571–6573; (g) Kumar, P.; Gupta, P.; Naidu, S. V. Chem.—Eur. J. **2006**, *12*, 1397–1402; (h) Kumar, P.; Naidu, S. V. J. Org. Chem. **2006**, *71*, 3935–3941.

- (a) Naidu, S. V.; Kumar, P. *Tetrahedron Lett.* 2003, 44, 1035–1037; (b) Kandula, S. V.; Kumar, P. *Tetrahedron Lett.* 2003, 44, 1957–1958; (c) Gupta, P.; Naidu, S. V.; Kumar, P. *Tetrahedron Lett.* 2004, 45, 9641–9643; (d) Kondekar, N. B.; Kandula, S. V.; Kumar, P. *Tetrahedron Lett.* 2004, 45, 5477–5479; (e) Pandey, S. K.; Kandula, S. V.; Kumar, P. *Tetrahedron Lett.* 2004, 45, 5877–5879; (f) Kandula, S. V.; Kumar, P. *Tetrahedron: Asymmetry* 2005, 16, 3579–3583.
- (a) Sonogashira, K.; Tohda, Y.; Hagihara, N. *Tetrahedron Lett.* 1975, 16, 4467–4470; (b) Yu, Q.; Wu, Y.; Ding, H.; Wu, Y.-L. *J. Chem. Soc., Perkin Trans. 1* 1999, 1183–1188; (c) Madec, D.; Férézou, J.-P. *Tetrahedron Lett.* 1997, 38, 6661–6664; (d) Izzo, I.; Decaro, S.; De Riccardis, F.; Spinella, A. *Tetrahedron Lett.* 2000, 41, 3975–3978.
- (a) Becker, H.; Sharpless, K. B. Angew. Chem., Int. Ed. 1996, 35, 448–451;
 (b) Kolb, H. C.; VanNieuwenhze, M. S.; Sharpless, K. B. Chem. Rev. 1994, 94, 2483–2547.
- (a) Mancuso, A. J.; Brownfain, D. S.; Swern, D. J. Org. Chem. 1979, 44, 4148–4150; (b) For reviews on the Swern oxidation, see: (i) Tidwell, T. T. Synthesis 1990, 857–870; (ii) Tidwell, T. T. Org. React. 1990, 39, 297–572.
- 25. (a) Yadav, J. S.; Chander, M. C.; Joshi, B. V. *Tetrahedron Lett.* 1988, 29, 2737–2740; (b) Takano, S.; Sugihara, T.; Ogasawara, K. *Heterocycles* 1990, 31, 1721–1725; (c) Takano, S.; Yoshimitsu, T.; Ogasawara, K. *Synlett* 1994, 119–120.
- Takai, K.; Nitta, K.; Utimoto, K. J. Am. Chem. Soc. 1986, 108, 7408–7410.
- 27. Schon, I. Chem. Rev. 1984, 84, 287-297.
- Frigerio, M.; Santagostino, M. *Tetrahedron Lett.* **1994**, *35*, 8019–8022.
- 29. Dalcanale, E.; Montanari, F. J. Org. Chem. 1986, 51, 567-569.
- Oikawa, Y.; Yoshioka, T.; Yonemitsu, O. *Tetrahedron Lett.* 1982, 23, 885–888.
- Compound registration no. H-02540, *Dictionary of Natural Products*; Buckingham, J., Ed.; Chapman and Hall: London, 1994; Vol. 3, p 3128.
- (a) Drouet, K. E.; Theodorakis, E. A. Chem.—Eur. J. 2000, 6, 1987–2001; (b) Smith, A. B., III; Ott, G. R. J. Am. Chem. Soc. 1998, 120, 3935–3948; (c) Smith, A. B., III; Wan, Z. J. Org. Chem. 2000, 65, 3738–3753.
- (a) Nicolaou, K. C.; Ladduwahetty, T.; Taffer, I. M.; Zipkin,
 R. E. Synthesis 1986, 344–347; (b) Overman, L. E.;
 Thompson, A. S. J. Am. Chem. Soc. 1988, 110, 2248–2256.